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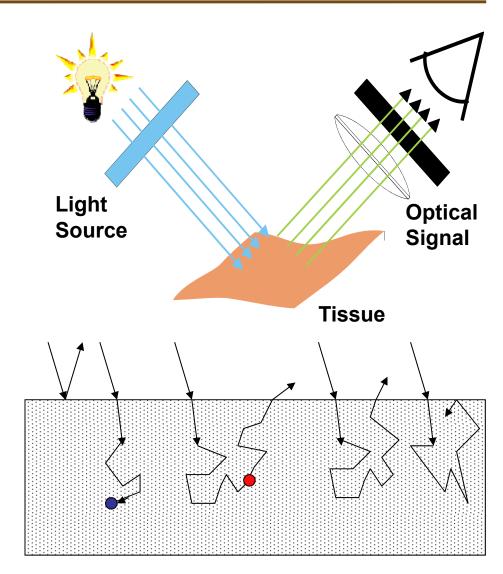
#### Introduction

#### Interaction between Light and Tissue

- Reflection
- Refraction
- Absorption
- Fluorescence
- Scattering

#### Depends on

- Constituents of tissue
- Optical properties of tissue
- Propagation of light





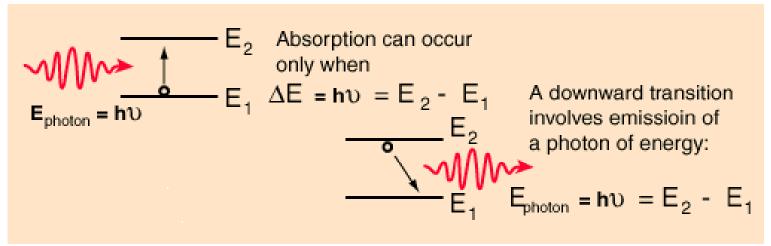




- Extraction of energy from light by a molecular species
- Diagnostic applications: Transitions between two energy levels of a molecule that are well defined at specific wavelengths could serve as spectral fingerprint of the molecule
  - Various types of Chromophores (light absorbers) in Tissue
  - Wavelength-dependent absorption
  - Tumor detection and other physiological assessments (e.g. pulseoximetry)
- Therapeutic applications: Absorption of energy is the primary mechanism that allows light form a source (laser) to produce physical effects on tissue for treatment purpose
  - Lasik (Laser Assisted in situ Keratomileusis) Eye Surgery, Tatoo Removal, PDT



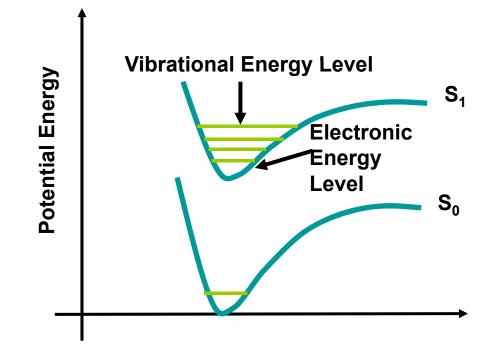
- Absorption occurs when the photon frequency matches the 'frequency' associated with the molecule's energy transition
  - Electrons absorb the energy of the light and transform it into vibrational motion
  - The absorption of a photon results in:
  - quantized change in charge separation
  - quantized excitation of vibrational modes
  - Electrons interact with neighboring atoms → convert vibrational energy into thermal energy



5

#### Absorption

- Each electronic energy levels is associated with many vibrational energy levels
- Absorption of UV and visible light promotes transition between electronic energy levels
- Absorption of infrared light promotes transitions between vibrational energy levels



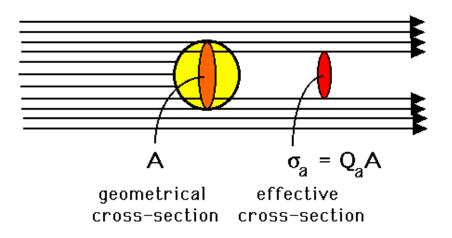


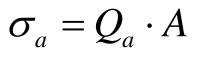
#### 6

# Absorption

# • Absorption Cross-section, $\sigma$ [m<sup>2</sup>]

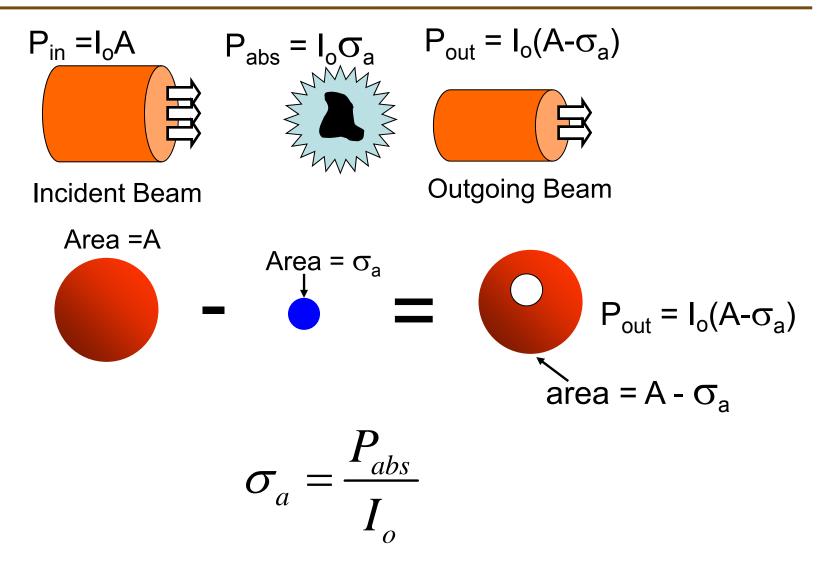
- Consider a chromophore idealized as a sphere with a particular geometrical size.
- Consider that this sphere blocks incident light and casts a shadow, which constitutes absorption.
- The size of absorption shadow
   = absorption cross-section
- Qa: absorption efficiency











7



#### Assumptions

- Cross section is independent of relative orientation of the impinging light and absorber uniform distribution of N<sub>a</sub> (molecules/cm<sup>3</sup>) identical absorbing particles
- Absorption Coefficient, ma [1/m]

$$\mu_a = N_a \cdot \sigma_a$$

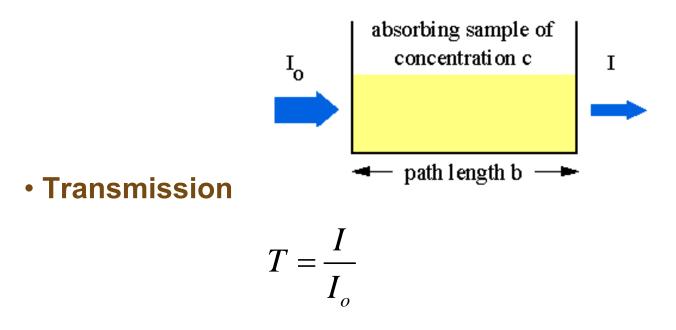
- Absorption cross-sectional area per unit volume of medium
- Absorption mean free path, la [m]

$$l_a = \frac{1}{\mu_a}$$

 Represents the average distance a photon travels before being absorbed



Transmission and Absorbance (macroscopic view)



Absorbance (attenuation, or optical density)

$$A = -\log(T) = \log\left(\frac{I_o}{I}\right)$$

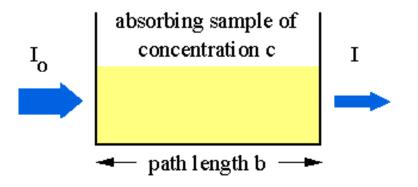


#### • Lambert – Beer Law:

- The linear relationship between absorbance and concentration of an absorbing species.
- Relates  $\mu_{\alpha},$  transmission, and absorbance

$$I = I_o e^{-\mu_a \cdot b}$$

$$\mu_a = N_a \cdot \sigma_a$$
$$\sigma_a = \frac{P_{abs}}{I_o}$$



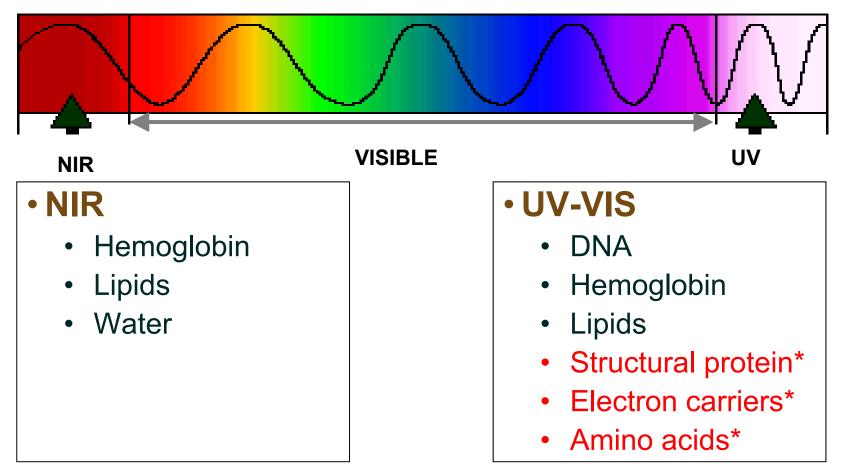
 $\sigma$ = absorption cross-sectional area [cm<sup>2</sup>] I<sub>o</sub> = The intensity entering the sample at z = 0 [w/cm<sup>2</sup>] I = The intensity of light leaving the sample [w/cm<sup>2</sup>]

b = pathlength traveled in the sample [cm]





#### **Absorbers in Tissue**

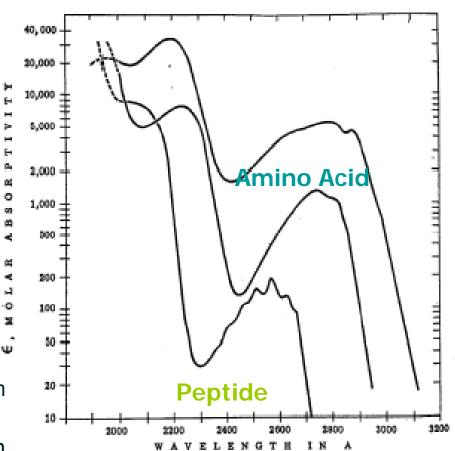


\* Absorbers that fluoresce when excited in the UV-VIS



#### **UV Absorption**

- Protein, amino acid, fatty acid and DNA absorption dominate UV absorption
- Protein
  - Dominant 'non-water' constituent of all soft tissue, ~ 30%
  - Absorption properties determined by peptide bonds and amino acid residues
    - Peptide excitation about λ = 190 nm
    - Amino acids absorption at λ = 210 - 220 nm and 260 – 280 nm
- DNA
  - Absorbs radiation for  $\lambda \leq 320$  nm
- Large water absorption λ< 180 nm</li>

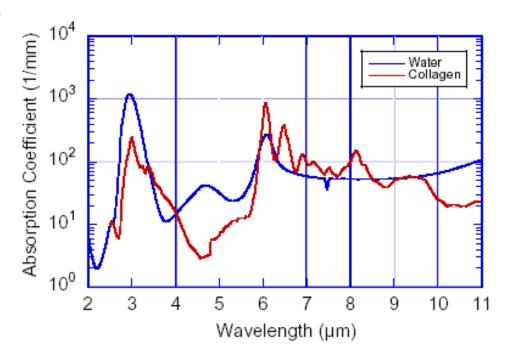


#### 13

## Absorption

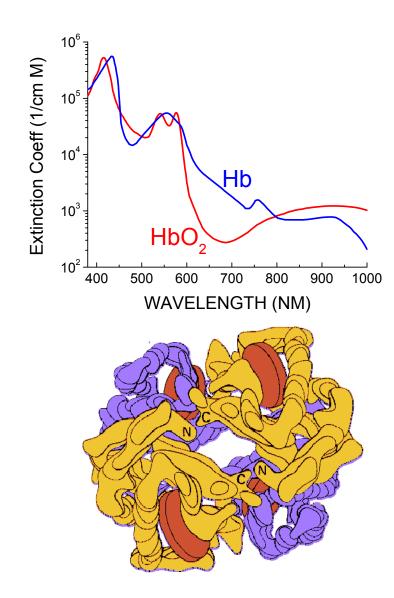
#### **Infrared Absorption**

- Protein IR absorption peaks at 6.1, 6.45, and 8.3 µm due to amide excitation
- Absorption depth  $\leq 10 \ \mu m$  in  $\lambda = 6-7 \ \mu m$  region
- Water absorption peak at 0.96, 1.44, 1.95, 2.94 and 6.1 μm
  - Absorption depth
    - ~ 500 mm at  $\lambda$  = 800 nm
    - <1 μm at λ=2.94 μm</li>
    - $\leq 20 \ \mu m$  throughout  $\lambda \geq 6 \ \mu m$









Main Absorbers at visible and NIR
➢ Hemoglobin
➢ Lipid

#### Hemoglobin

- Each hemoglobin has 4 heme (Fe<sup>2+</sup>) sites to bind O<sub>2</sub>
- Responsible for oxygen transport
  HbO<sub>2</sub> and Hb
- oxygen saturation is an indicator of oxygen delivery and utilization as well as metabolic activity



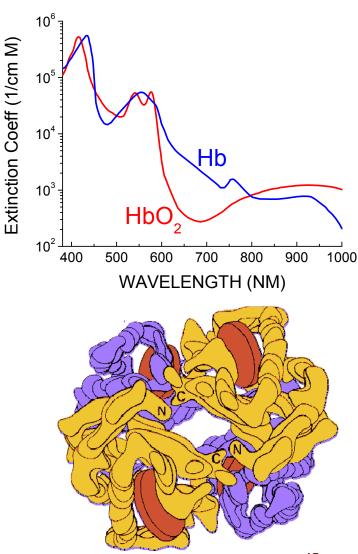
#### Hemoglobin

#### Responsible for oxygen transport

- HbO2 and Hb
- oxygen saturation is an indicator of oxygen delivery and utilization as well as metabolic activity
- Deoxyhemoglobin has lower absorption than oxyhemoglobin in the blue and green
  - Bright red arterial blood
  - Bluish venous blood

#### Absorption peaks for HbO<sub>2</sub>

- 418, 542, 577, and 925 nm
- Absorption peaks for Hb
  - 550, 758, 910 nm
- Isosbestic points
  - 547, 569, 586, and 798 nm

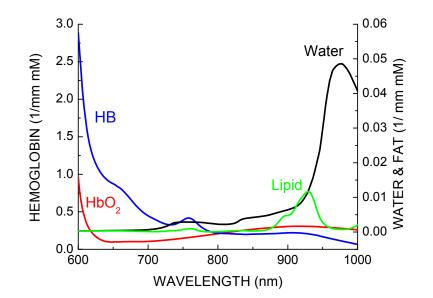


#### 16

## Absorption

#### Lipid (Fat)

- Important energy store in the body
- Site-specific measurements of body composition
- Monitoring of physiological changes in female breast tissue
- Tissue layer model







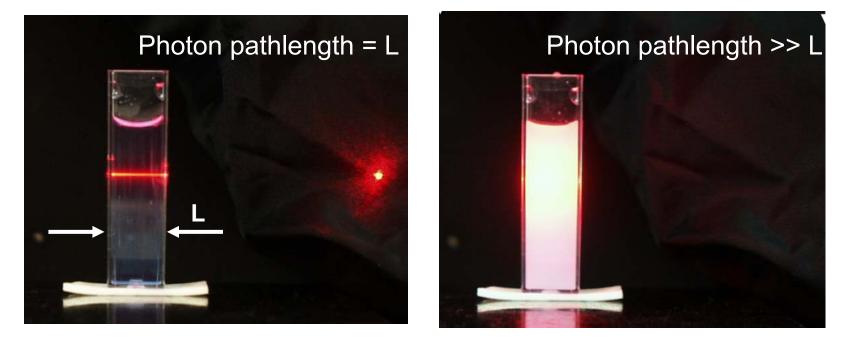


- Change of direction of propagation and/or energy of light by a molecular species
- Diagnostic applications: Scattering depends on the size, morphology, and structure of the components in tissues (e.g. lipid membrane, collagen fibers, nuclei).
  - Variations in these components due to disease would affect scattering properties, thus providing a means for diagnostic purpose
- Therapeutic applications: Scattering signals can be used to determine optimal light dosimetry and provide useful feedback during therapy



#### Purely absorbing

#### With Scattering



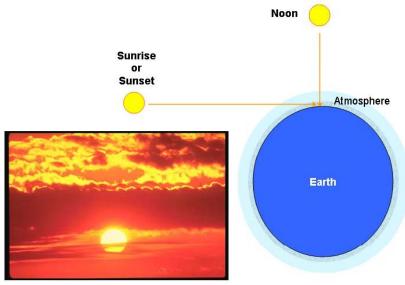
Lambert- Beer Law does not apply here!!! Need to calculate true pathlength of light



# • Why is the sky blue, clouds white, and sunsets red?

- Blue skies are produced due to scattering at shorter wavelengths
  - Visible light (violet & blue) are selectively scattered by O2 and N2 – much smaller than wavelengths of the light
  - violet and blue light has been scattered over and over again
- When light encounters larger particles (cloud, fog), Mie scattering occurs
  - Mie scattering is not wavelength dependent appears white
  - Cigarette smoke, too
- At sunset
  - The light must travel over a longer path in the atmosphere
  - Blue/green is scattered away and only red/orange (scattered less) reaches your eyes







#### Mechanism for Light Scattering

- Light scattering arises from the presence of heterogeneities within a bulk medium
  - Physical inclusions
  - Fluctuations in dielectric constant from random thermal motion
- Heterogeneity/fluctuations → non-uniform temporal/spatial distribution of refractive index in the medium
  - Passage of an incident EM wave sets electric charges into oscillatory motion and can excite vibrational modes
  - Scattered light is re-radiated by acceleration of these charges and/or relaxation of vibrational transition





#### Elastic scattering: no energy change

- Frequency of the scattered wave = frequency of incident wave
- Probes static structure of material
- Rayleigh and Mie scattering

#### Inelastic scattering: energy change

- Frequency of the scattered wave ≠ frequency of incident wave
- Internal energy levels of atoms and molecules are excited
- Probes vibrational bonds of the molecule
- Raman scattering (stokes↓ and anti-stokes ↑)

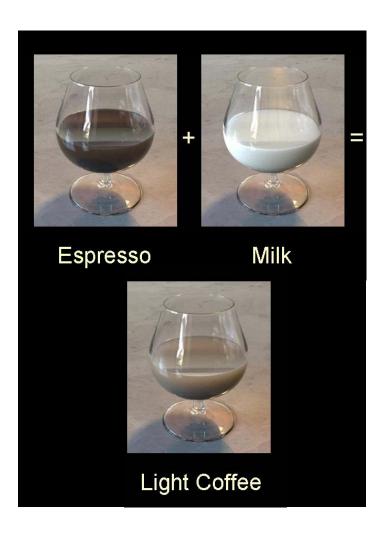
#### 22

#### Scattering

#### **Elastic Scattering**

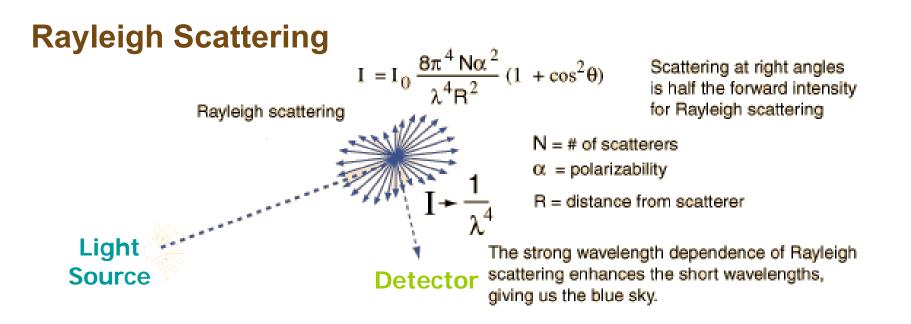
- The light scattered by a system has interacted with the inhomogeneities of the system
- Photons are mostly scattered by the structure whose size matches the wavelength
- Principal parameters that affect scattering
  - Wavelength,  $\lambda$
  - Relative refractive index
  - Particle radius
  - Shape and orientation

#### • Two types of scattering: Rayleigh and Mie









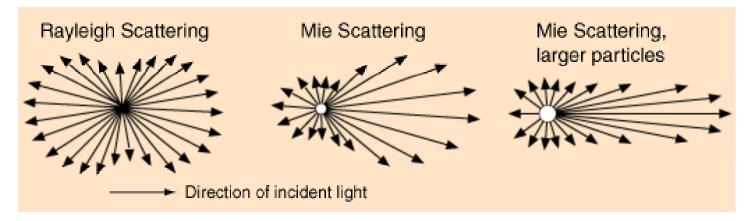
- Scattering from very small particles  $\rightarrow \leq \lambda/10$
- Rayleigh scattering is inversely related to fourth power of the wavelength of the incident light

$$I \propto \frac{1}{\lambda^4}$$

 $\lambda$  is the wavelength of the incident light I is the intensity of the scattered light

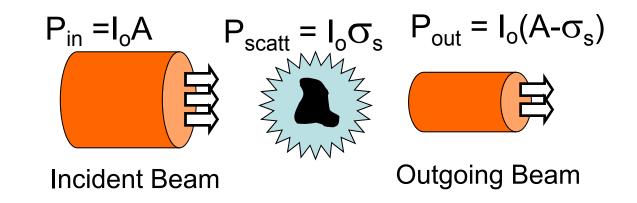


#### **Mie Scattering**



- For scattering of particles comparable or larger than the wavelength, Mie scattering predominates
- Because of the relative particle size, Mie scattering is not strongly wavelength dependent
- Forward directional scattering





- Scattering Cross Section, σ<sub>scatt</sub> [m<sup>2</sup>]
  - 'area' of an index-matched, perfectly absorbing disc necessary to produce
- The measured reduction of light

 $\sigma_{scatt} = Q_s * A_s$ 

- Q<sub>s</sub>: Scattering efficiency (calculated by Mie theory); defined as the ratio of the scattering section to the projected area of the particle on the detector
- A<sub>s</sub>: Area of Scatterer [m2]



#### • Scattering Coefficient, $\mu_s$ [1/m]

- $\mu_s = N_s \sigma_s$  ,
  - $N_s$  = the number density of scatterers
  - $\sigma_s$  = scattering efficiency
- Cross-sectional area for scattering per unit volume of medium
- Scattering Mean Free Path, I<sub>s</sub>
  - Average distance a photon travels between scattering events

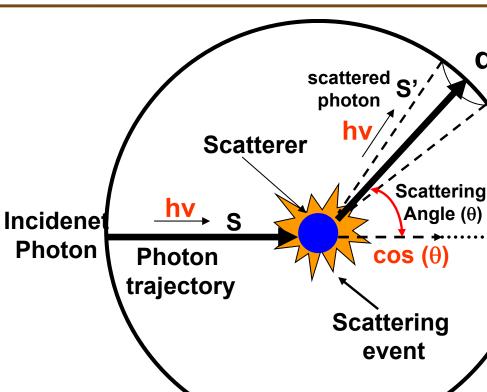
$$l_s = \frac{1}{\mu_s}$$



 $d\Omega$ 

#### • Anisotropy, g

- Imagine that a photon is scattered by a particle so that its trajectory is deflected by an angle, θ
- Then, component of a new trajectory aligned forward direction is cos(θ)
- Anisotropy is a measure of forward direction retained after a single scattering event, < cos(θ)>

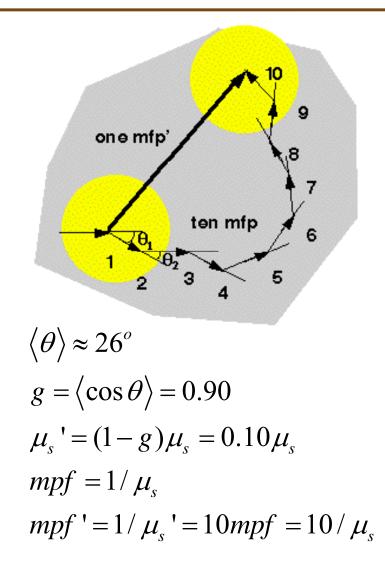


- $\left(-1 \text{ totally backward scattering}\right)$
- $g = \begin{cases} 0 & \text{isotropic scattering} \end{cases}$ 
  - totally forward scattering

Biological Tissues: 0.65 < g < 0.95



- Reduced Scattering Coefficient, µs' [1/m]
  - µs' incorporates the scattering coefficient, µs and the anisotropy factor, g
    - $\mu_{\rm s}' = (1-g)\mu_{\rm s}$
  - µs' can be regarded as an effective isotropic scattering coefficient that represent the cumulative effect of several forwardscattering events
  - Special significant with respect to photon diffusion theory





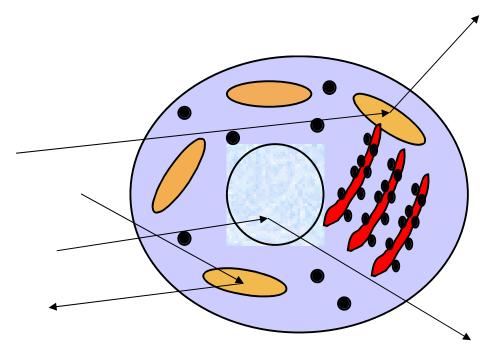
# Scattering in Tissue Tissue is composed of a 'mixture' of Rayleigh and Mie scattering 10 μm cells nuclei mitochondria

**1** μ**m** lysosomes, vesicles Mie Scattering striations in collagen fibrils **0.1** μm macromolecular aggreagates **Rayleigh Scattering** membranes **0.01** μm · 29



#### Scattering in Tissue

- Refractive index mismatch between lipid and surrounding aqueous medium
  - Soft tissues are dominated by lipid contents
  - Celluar membranes, membrane folds, and membraneous structure
- Mitochondria, ~ 1μm
  - Intracelluar organelle composed of many folded membrane, cristae
- Collegan fibers, 2 ~ 3μm
  - Collegan fibrils, 0.3 μm
  - Periodic fluctuation in collegan ultrastructure →source of Rayleigh scattering in UV and Visible range
- Cells

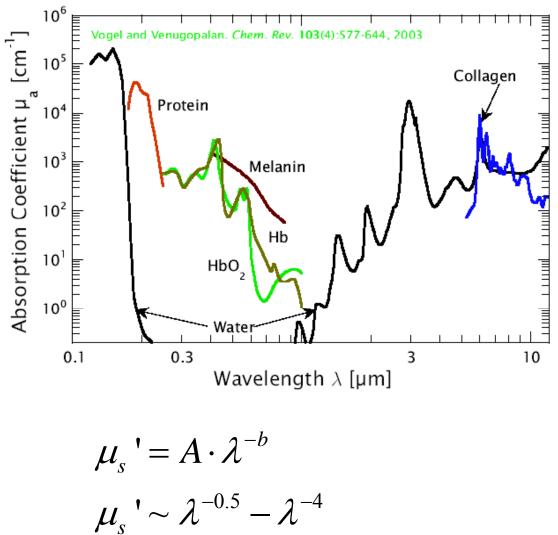




 Scattering and absorption occur simultaneously and are wavelength dependent

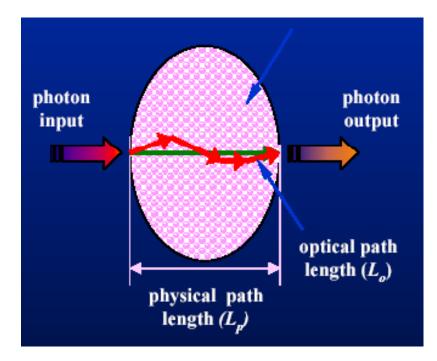
$$\mu_t = \mu_s + \mu_a$$

- Scattering monotonically decreases with wavelength
- Absorption is large in UV, near visible, and IR
- Absorption is low in red and NIR → Therapeutic window (600 ≤λ≤ 1000 nm)





- Modeling of light transport in tissues are often governed by the relative magnitudes of optical absorption and scattering
  - µa >> µs' : Lambert-Beer Law (λ ≤300nm;λ≥2000nm)
  - μs' >> μa : Diffusion
     Approximation (600nm ~ 1000nm)
  - μs' ~ μa : Equation of Radiative Transfer, Monte Carlo (300nm ~ 600 nm; 1000nm ~ 2000nm)
- Use Monte Carlo, Transport Theory, or Diffusion Theory



Physical Pathlength: L<sub>p</sub> Optical Pathlength: L<sub>o</sub>

Biological Tissue Lo/L<sub>p</sub> = 4 or ↑

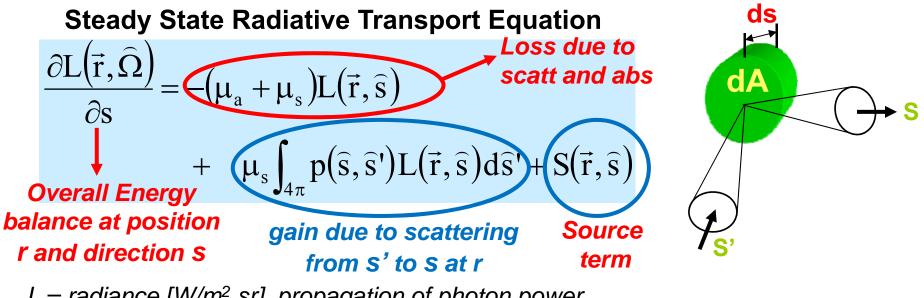


 Modeling Photon Propagation Random Walk depiction of photon propagation in a input homogeneous medium. This detector comprises of combinations of multiple-scatter, absorption and detection. photon photon detected **Monte Carlo** absorbed  $\mu_a$ ,  $\mu_s$ , g, phase function S "Stochastic" Description



#### Radiative Transport Theory

- The direct application of EM theory is complicated
- RTT deals with the transport of light energy
- RTT ignores wave phenomena (polarization, interference) of EMT



- $L = radiance [W/m^2 sr]$ , propagation of photon power
- *P*(s, s') = phase (scattering) function
- s, s' = directional vectors of photon propagation



#### Diffusion Approximation

- Simplified form of RTT at "diffusion limit"
- µs' >> µa
  - the number of photon undergoing the random walk is large

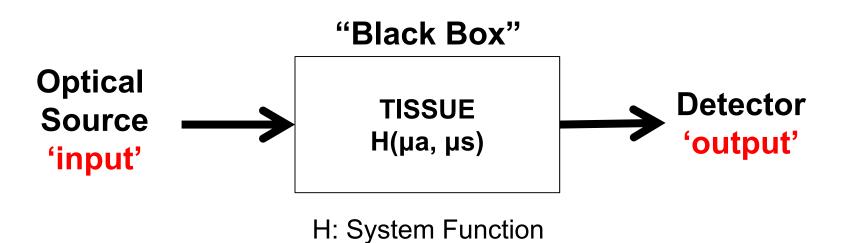
$$\partial j(\vec{r},t) / \partial t \ll c(\mu_a + \mu_s') = c\mu_t'$$
$$L(\vec{r},\hat{s},t) \approx \frac{1}{4\pi} \phi(\vec{r},t) + \frac{3}{4\pi} \vec{j}(\vec{r},t) \cdot \hat{s}$$

- Isotropic source beyond 1/µt'
  - ~10/µť (~ 1mm in biological tissue)
  - far from sources & boundaries
  - assume tissue is "macroscopically homogeneous"

$$\frac{1}{c} \frac{\partial \phi(\vec{r},t)}{\partial t} - \nabla \cdot D(\vec{r}) \nabla \phi(\vec{r}) - \mu_a \phi(\vec{r},t) = S(\vec{r},t)$$
  
where  $D(\vec{r}) = 1/3 [\mu_a(\vec{r}) + \mu_s(\vec{r})]$ 



#### Measurement Strategies



- Goal: To find out H(µa, µs)
- Requires Non-Static system → Perturbations in either optical source or tissue



#### **Measurement Schemes**

- CW (Continuous Wave) Measurement
  - Simplest form of measurement
  - Static, continuous wave input
  - requires dynamic tissue property changes
  - E.g. pulse oximetery

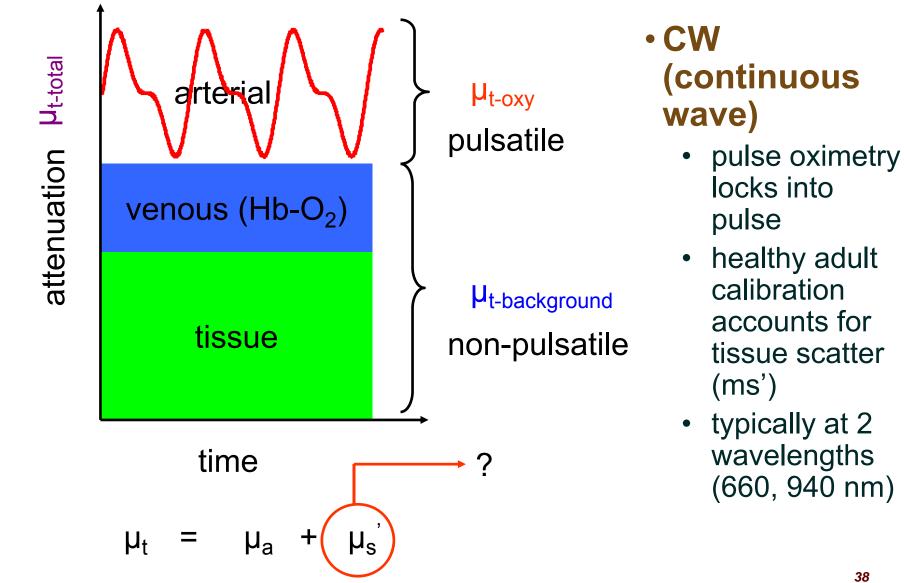
#### Time-Resolved Measurements

- Temporal changes in optical sources
- Time Domain Photon Migration (TDPM)
- Frequency Domain Photon Migration (FDPM)

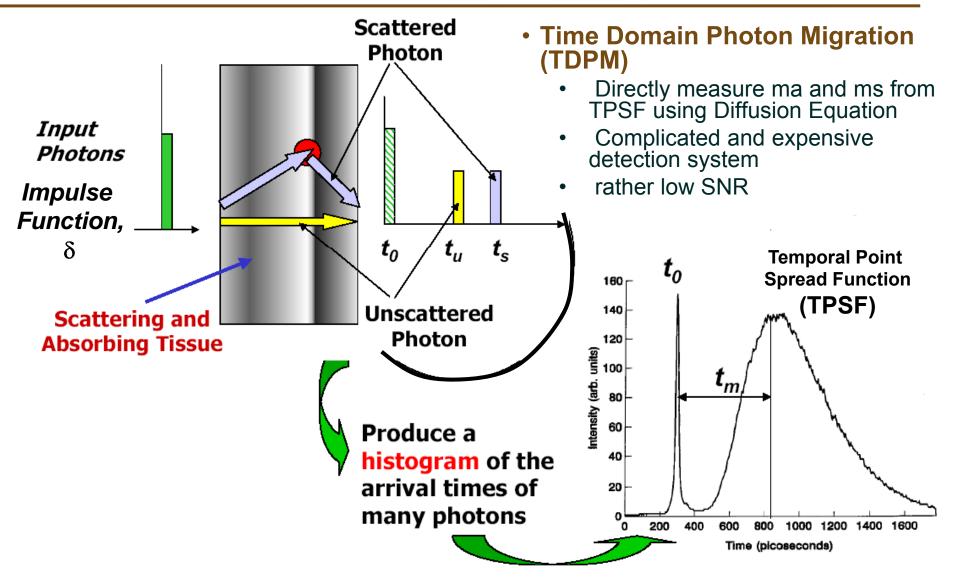
#### Spatially-Resolved Measurement

• Spatial changes in optical path











• Frequency Domain Photon Migration (FDPM)

